

## Descriptions — Chemistry

### of Courses

## CHEMISTRY CEM

### College of Natural Science

Credit cannot be earned in more than one course of each of the following groups: 141 and 151; 143, 241, and 351; 142 and 153; 242 and 352; 243 and 354; 244 and 355; 245 and 353; 361 and 383; 363 and 385; 384 and 461.

With department approval, students with advanced placement credit in CEM 151 and 161 may enroll in CEM 181H and 184H. Those with advanced placement credit in CEM 152 may enroll in CEM 182H, and those with advanced placement credit in CEM 153 may enroll in CEM 183H. CEM 181H-182H-183H is a more advanced treatment of material in CEM 151-152-153. CEM 184H-185H-186H is a more advanced treatment of material in CEM 161-162-163. Students with credit in an honors chemistry course may not enroll in the corresponding nonhonors course.

#### 139. Selected Topics in Introductory Chemistry

Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 7 credits. Previous college chemistry, approval of department.

Self-instructional units from CEM 140, CEM 141A, CEM 141B (or equivalent) selected and approved by the department for individual students with special needs.

#### 140. Introductory Chemistry

Fall, Winter, Spring, Summer. 2 credits. Self-scheduled instruction only. MTH 108 or MTH 111 or concurrently.

Chemical symbols, formulas, equations, stoichiometry, structure of atoms, bonding, states of matter, solutions.

#### 141. Chemical Principles

(CEM 141A, CEM 141B.) Fall, Winter, Spring, Summer. 4(4-0) MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.

Chemical principles: stoichiometry, mole concept, atomic and molecular structures, states of matter, gases, crystal structure, solutions, kinetics, thermodynamics, acids and bases, aqueous equilibria.

#### 141B. Chemical Principles

Fall, Winter, Spring, Summer. Fall 4(4-0); Winter, Spring, Summer: 4 credits. Self-scheduled instruction only. MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.

Chemical principles for students in biological, health-related, and agricultural disciplines. Approved through Fall 1989.

#### 142. Descriptive Inorganic Chemistry

Winter, Spring. 3(3-0) CEM 141A or CEM 141B or CEM 152.

Reactions and behavior of inorganic compounds illustrated in part by industrial and environmental applications.

#### 143. Introductory Organic Chemistry

Fall, Spring, Summer. 4(3-3) CEM 141A or CEM 141B or CEM 152.

Chemistry of carbon compounds. The chemistry of the main organic functional groups is described and illustrated with applications to everyday life, industry and biology. The laboratory is designed to exemplify the lectures.

#### 151. Principles of Chemistry I

Fall, Winter. 4(4-0) MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.

First of a 3-term sequence for science majors, chemical engineering students, and others desiring a comprehensive general chemistry sequence. Atomic and molecular structure; stoichiometry; solids, liquids, and gases; solutions.

#### 152. Principles of Chemistry II

Winter, Spring. 3(3-0) MTH 112 or concurrently; CEM 151 or CEM 141A or CEM 141B or CEM 181H.

Continuation of CEM 151. Chemical thermodynamics; kinetics, acids, bases, and aqueous equilibria; electrochemistry.

#### 153. Introductory Inorganic Chemistry

Fall, Spring. 3(3-0) CEM 152 or CEM 182H.

Continuation of CEM 152. Descriptive inorganic chemistry with further discussion of bonding.

#### 161. Introductory Chemistry Laboratory

Fall, Winter, Spring, Summer. 1(0-3) CEM 140 or CEM 141A or CEM 141B or CEM 151 or concurrently.

Laboratory work in chemistry, including quantitative physicochemical or analytical experiments and chemical synthesis.

#### 162. Quantitative Analysis

Fall, Winter, Spring, Summer. 3(1-6) CEM 141A or CEM 141B or CEM 151 or CEM 181H; CEM 161 or CEM 184H.

Laboratory work in quantitative chemistry.

#### 163. Introductory Inorganic Laboratory

Spring. 2(0-6) CEM 142 or CEM 153 or concurrently; CEM 161.

Qualitative analysis and inorganic preparations.

#### 181H. Honors Chemistry I—Principles

Fall. 4(4-0) An A average in high school chemistry, physics and mathematics; MTH 112 or MTH 122 concurrently. Results of examination during orientation; approval of department.

Subatomic, atomic and molecular structure; quantum theory and bonding; experimental methods of structure determination; states of matter; nuclear chemistry.

#### 182H. Honors Chemistry II—Principles

Winter. 4(4-0) CEM 181H with grade of 3.0 or better and/or approval of department. MTH 113 or MTH 123 concurrently.

Kinetic theory of gases, thermodynamics, chemical equilibrium, electrochemistry, chemical kinetics, properties of solutions, macromolecular chemistry.

#### 183H. Honors Chemistry III—Inorganic Chemistry

Spring. 3(3-0) CEM 182H with grade of 3.0 or better and/or approval of department.

Descriptive inorganic chemistry by periodic groups of elements. Nomenclature, bonding, stereochemistry, and reactions of compounds of the representative and transition elements.

#### 184H. Honors Chemistry Laboratory I

Fall. 1(0-3) CEM 181H concurrently; approval of department.

Techniques of measurement; errors and significant figures; experiments related to atomic and molecular structure.

#### 185H. Honors Chemistry Laboratory II

Winter. 2(0-6) CEM 184H; CEM 182H concurrently; approval of department.

Experiments related to gas behavior, thermodynamics, electro-chemistry, chemical kinetics and properties of solutions.

#### 186H. Honors Chemistry Laboratory III

Spring. 2(0-6) Approval of department.

Introductory independent laboratory work in chemistry.

#### 241. Organic Chemistry

Fall, Winter, Summer. 4(4-0) CEM 141A or CEM 141B or CEM 152 or CEM 181H; CEM 161 or CEM 184H.

Common classes of organic compounds with emphasis on nomenclature, structural principles, reactions and reaction mechanisms.

#### 242. Organic Chemistry

Winter, Spring, Summer. 4(4-0) CEM 241.

Continuation of CEM 241 with emphasis on polyfunctional compounds, particularly groups of compounds having biological significance.

#### 243. Organic Chemistry Laboratory

Fall, Winter: 1(0-2) Summer: 1(0-3). CEM 241 or concurrently.

Introduction to standard organic laboratory techniques.

#### 244. Organic Chemistry Laboratory

Winter, Spring, Summer. 1(0-3) CEM 241, CEM 243, CEM 242 concurrently.

Organic preparations and qualitative analysis.

#### 245. Organic Chemistry

Fall, Spring. 4(4-0) CEM 242.

Selected topics of organic chemistry, especially compounds of biological interest, discussed with emphasis on mechanisms and stereochemistry. Topics include polymers, amino acids, proteins, sugars, terpenes, steroids, and alkaloids.

#### 333. Instrumental Methods

Spring. 4(2-6) CEM 143 or CEM 241 or CEM 351; CEM 162.

Principles, applications of separation and instrumental analysis. Atomic emission, absorption, fluorescence spectrometry; UV, visible, IR spectrophotometry; molecular fluorescence; gas and other chromatography; electro-analytical chemistry; electrophoresis; radiochemistry.

#### 351. Organic Chemistry

Fall. 3(4-0) CEM 152 or CEM 182H.

A comprehensive introduction to the fundamentals of organic chemistry designed for chemistry majors but open to others who desire a rigorous, modern treatment of the subject.

#### 352. Organic Chemistry

Winter. 3(4-0) CEM 351.

Continuation of CEM 351.

#### 353. Organic Chemistry

Spring. 3(4-0) CEM 352.

Continuation of CEM 352.

#### 354. Organic Chemistry Laboratory

Winter. 2(0-6) CEM 162 or CEM 185H; CEM 241 or CEM 351.

Introduction to modern organic laboratory techniques.

- 355. Organic Chemistry Laboratory**  
Spring. 2(0-6) CEM 352, CEM 354.  
Application of modern organic laboratory techniques to organic synthesis and analysis.
- 356. Organic Chemistry Laboratory**  
Fall. 2(0-6) CEM 355.  
Advanced organic syntheses, qualitative organic analysis, and practical literature searching.
- 361. Chemical Thermodynamics**  
Fall. 3(4-0) *One year general chemistry; one year general physics; MTH 215.*  
Thermodynamics. Properties of gases. Laws of thermodynamics, properties of ideal and non-ideal solutions, thermodynamics of chemical reactions, activities in non-ionic systems.
- 362. Analytical-Physical Chemistry I**  
Winter. 3(4-0) CEM 361.  
Applications of thermodynamics. Activity coefficients, ionic solutions, cell potentials, ionic equilibria including acid-base, complexation, solubility and redox equilibria, phase equilibria, distillation, extraction, chromatography.
- 363. Analytical-Physical Chemistry II**  
Spring. 3(4-0) CEM 362.  
Chemical kinetics. Homogeneous kinetics, reaction mechanisms, temperature dependence of reaction rates, transport process, heterogeneous kinetics, electrode kinetics, X-ray diffraction, crystal structure.
- 372. Analytical-Physical Chemistry Laboratory I**  
Winter. 2(1-3) CEM 162; CEM 383, or CEM 361.  
Electronics and chemical instrumentation; spectroscopy, optical methods and lasers.
- 373. Analytical-Physical Chemistry Laboratory II**  
Spring. 2(1-3) CEM 372.  
Instrumental measurements. Electrode potentials, chromatography, spectrophotometry, electrolytic conductance, solution kinetics.
- 383. Physical Chemistry: Introductory**  
Fall, Summer. 3(4-0) CEM 143 or CEM 241 or CEM 351; MTH 113.  
Classical and chemical thermodynamics. Introduction to the laws and their applications in treating chemical reactions, pure substances, ideal and non-ideal mixtures, and colligative properties.
- 384. Physical Chemistry: Introductory**  
Spring. 3(4-0) CEM 383.  
Atomic and molecular structure. Atomic and molecular orbitals and chemical bonding. Rotational, vibrational and electronic spectra, nuclear magnetic resonance and electron spin resonance.
- 385. Physical Chemistry: Introductory**  
Winter. 3(4-0) CEM 383.  
Electrochemistry and electromotive force. Chemical kinetics. Macromolecules and biochemical systems. Nuclear chemistry.
- 400H. Honors Work**  
Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 18 credits. Seniors, approval of department.  
Assigned reading and investigation in chemistry under the supervision of the staff. The program will include some creative work.
- 411. Inorganic Chemistry I**  
Fall. 3(3-0) CEM 385 or CEM 363.  
Principles of structure and bonding in inorganic chemistry, crystal symmetry, coordination chemistry, solvent systems, hydrogen bonding and selected examples from the chemistry of various elements.
- 412. Inorganic Chemistry II**  
Winter. 3(3-0) CEM 411.  
Inorganic chemistry viewed in a variety of examples of reactions, structure, mechanisms, etc., from representative main group elements and transition elements.
- 419. Problems and Reports**  
Fall, Winter, Spring, Summer. 1 to 8 credits. May reenroll for a maximum of 8 credits. Students may not reenroll if they have an outstanding DF in CEM 419. Approval of department.
- 430. Introduction to Radioactivity and Radioisotope Techniques**  
(PHY 430.) Spring. 3(3-0) *One year each of general college chemistry and physics.*  
Elementary nuclear processes and properties with emphasis on radioactivity, its measurement, and its interaction with matter. Effects of radiation on chemical and biological systems. Applications of nuclear technology, safety and environmental factors. Fundamentals of nuclear models, reactions and decay mechanisms. Basic principles of nuclear reactors and accelerators.
- 431. Laboratory for Radioactivity and Radioisotope Techniques**  
(PHY 431.) Spring. 1(0-3) CEM 161, PHY 430 concurrently. CEM 162 recommended.  
Introduction to nuclear instrumentation. Experimental techniques for application of radioisotopes to problems in chemistry, the life sciences, and industry.
- 461. Theoretical Chemistry I**  
Fall. 3(4-0) *One year general chemistry; one year general physics; MTH 215.*  
Quantum chemistry. Wave properties, postulates of quantum mechanics, hydrogen atom, helium atom, orbital theories, ionic bonds, simple molecules, valence-bond and molecular-orbital theories, complex molecules.
- 462. Theoretical Chemistry II**  
Winter. 3(4-0) CEM 361, CEM 461.  
Spectroscopy and molecular structure. Electronic, infrared, Raman, and microwave spectroscopy, magnetic susceptibility and magnetic resonance, statistical mechanics, statistical thermodynamics, kinetic theory of gases, absolute rate theory.
- 471. Analytical-Physical Chemistry Laboratory III**  
Fall. 2(0-6) CEM 363, CEM 373.  
Kinetics, operational amplifiers, polarography, coulometry, electrochemical kinetics, stopped-flow kinetics, digital measurements, neutron activation.
- 472. Analytical-Physical Chemistry Laboratory IV**  
Winter. 2(0-6) CEM 461, CEM 471.  
Molecular properties. Mass spectrometry, nuclear and electron spin resonance spectroscopy, infrared Raman, and microwave spectroscopy, dipole moments, magnetic susceptibility.
- 499. Seminar on Chemical Physics**  
Fall, Winter, Spring. 1(1-0) *May reenroll for a maximum of 3 credits. One year of analytical-physical chemistry. MTH 215; PHY 428.*  
Literature of chemical physics through oral reports on selected journal articles in the area.
- 810. Advanced Inorganic Chemistry**  
Fall. 3(3-0) *Approval of department.*  
Structure, bonding, and reactivity patterns of inorganic compounds, with emphasis on non-metallic elements and reactions in nonaqueous media.
- 811. Symmetry, Group and MO Theory**  
Winter. 3(3-0) *Approval of department.*  
Applications of group and molecular orbital theory to chemical bonding, structure and reactions.
- 813. Advanced Inorganic Chemistry—Metals**  
Spring. 3(3-0) CEM 811.  
Continuation of CEM 811 with emphasis on the structure and chemistry of the metals.
- 833. Analytical Spectroscopy**  
Spring. 3(3-0) *Approval of department.*
- 834. Advanced Analytical Chemistry**  
Fall. 3(3-0) *Approval of department.*  
Consideration of principles and equilibria pertaining to aqueous and non-aqueous neutralization, redox and complexation reactions and the various separation techniques employed in analyses.
- 835. Spectrochemical Methods of Analysis**  
Winter. 3(3-0) or 4(3-4) *Approval of department.*  
Principles and applications of atomic absorption, emission, fluorescence; arc and spark emission spectroscopy; UV, visible, IR spectrophotometry; spectrophotometric titrations; reaction rate methods; molecular fluorescence, phosphorescence spectrometry; other optical spectrometric methods.
- 836. Separations**  
Spring of odd-numbered years. 3(3-0) *Approval of department.*  
Physical and chemical methods of separation.
- 837. Electroanalytical Chemistry**  
Spring of even-numbered years. 3(3-0) *Approval of department.*  
Theory and applications of modern electroanalytical chemistry to chemical and biomedical problems. Coulometry, electrometric titrations, ion-selective voltammetry; electrochemical synthesis and preparation of species for spectroscopy; trace analysis.
- 838. Scientific Instrumentation**  
Fall, Spring, Summer. 3(1-6) *May reenroll for a maximum of 9 credits. Approval of department.*  
Scientific measurements. Principles and applications of servo systems, operational amplifiers, linear and digital solid state devices, analog, digital and hybrid instrumentation systems, and minicomputers for scientific measurements.

## Descriptions — Chemistry

of

### Courses

#### 844. Structural Elucidation by Instrumental Methods

Fall. 3(3-0) Approval of department.

A practical instrumental analysis course with the major emphasis on the interpretation of data rather than a detailed description of the instrumentation. The fundamental principles behind the various measurements will be discussed in a general way, and important instrumental limitations will be noted.

#### 860. Organic Reactions: A Mechanistic Approach

Fall. 3(3-0) CEM 353; CEM 462 or approval of department.

Organic reactions are presented in a mechanistic framework. Reactions which proceed via carbonations, carbanions, free radicals, carbenes, arynes and other reactive intermediates, and concerted reactions are included.

#### 861. Structure of Organic Compounds

Winter. 3(3-0) CEM 860 or approval of department.

Structural and stereochemical principles will be developed and illustrated. Spectroscopic data will be used to illustrate the principles and to determine structure, with an emphasis on nuclear magnetic resonance spectroscopy.

#### 862. Advanced Synthetic Organic Chemistry

Spring. 3(3-0) CEM 860 or approval of department.

The strategy and methods of organic synthesis will be discussed.

#### 880. Atomic and Molecular Structure

Fall. 3(3-0) Approval of department.

Basic concepts of non-relativistic quantum mechanics will be developed and employed in a description of atomic and molecular structure.

#### 881. Thermodynamics

Winter. 3(3-0) Approval of department.

Laws of thermodynamics and their application to pure substances and solutions.

#### 883. Chemical Kinetics

Spring. 3(3-0) CEM 880.

Rates and mechanisms of chemical reactions, reaction rate theory, kinetic theory of gases, photochemistry.

#### 890. Graduate Problems and Reports

Fall, Winter, Spring, Summer. 1 to 6 credits. May reenroll for a maximum of 12 credits. Approval of department.

#### 899. Master's Thesis Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

Research in inorganic, analytical, organic, and physical chemistry.

#### 913. Selected Topics in Inorganic Chemistry

Fall, Spring, 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken.

Rare earth elements, recent advances in the chemistry of metals or nonmetals, high-temperature chemistry. Coordination chemistry and nonaqueous solvents.

#### 918. Seminar in Inorganic Chemistry

Fall, Winter, Spring. 1(2-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

#### 924. Selected Topics in Analytical Chemistry

Fall, Winter, Spring. 3(3-0) or 2(2-0) May reenroll for a maximum of 9 credits if different topic is taken.

Among topics which may be discussed are: advances in electroanalytical chemistry or spectroscopy; nonaqueous solvents; complexation equilibria; surface chemistry; analytical chemistry of polymers.

#### 938. Seminar in Analytical Chemistry

Fall, Winter, Spring. 1(1-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

#### 956. Selected Topics in Organic Chemistry

Fall, Winter, Spring. 2(2-0) or 3(3-0) May reenroll for a maximum of 12 credits if different topic is taken. Approval of department.

Topics may be selected from heterocyclic chemistry, natural products, free radicals, carbonium ions, organic sulfur or nitrogen compounds, acidity functions, isotope effects, photochemistry and others.

#### 958. Seminar in Organic Chemistry

Fall, Winter, Spring. 1(2-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

#### 985. Statistical Thermodynamics

Winter, Spring. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Partition functions, spectroscopic measurements and thermodynamic applications. Nonequilibrium statistical mechanics and thermodynamics. Time correlation functions and spectroscopic lineshapes, light scattering, and magnetic relaxation. Transport properties of fluids and gases.

#### 987. Selected Topics in Physical Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Mathematical preparation for quantum chemistry. Selected topics as: kinetics and photochemistry, macromolecular and surface chemistry, molecular spectroscopy, electro and magnetic properties of matter, application of statistical mechanics to chemical problems.

#### 988. Selected Topics in Physical Chemistry

Winter, Spring. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Topics may be chosen from analysis and interpretation of the spectra of molecules, advanced molecular structure, magnetic resonance, spectroscopy, X-rays and crystal structure, statistical mechanics.

#### 991. Selected Topics in Quantum Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Principles of quantum mechanics and application to chemical problems. Selected topics from spectroscopy, properties of atoms and molecules in electric and magnetic fields, and theories of molecular electronic structure.

#### 998. Seminar in Physical Chemistry

Fall, Winter, Spring. 1(1-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

#### 999. Doctoral Dissertation Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

Research in analytical, inorganic, organic, and physical chemistry.

## CHINESE

See Linguistics and Germanic, Slavic, Asian and African Languages.

## CIVIL AND ENVIRONMENTAL ENGINEERING

### College of Engineering

#### Civil Engineering

CE

#### 251. Elementary Surveying

Spring. 4(3-3) Not open to majors.

Use of the tape, compass, level, and transit with simple maps; traverse closure and area computations. Profile, cross section and stadia surveys, U.S. land system.

#### 252. Surveying I

Fall, Spring. 5(4-3) Engineering majors or approval of department.

Instruments, theory of measurements, error analysis, stadia, horizontal and vertical curves, U.S. Public Land System, observation for meridian.

#### 280. Introduction to Environmental Engineering

Fall, Spring. 4(4-0) CEM 141A, or CEM 151, MTH 112, CPS 112 or CPS 251.

Hydrology; ground water and surface water supply systems; wastewater treatment, methods of pollution control for solid waste, air, and noise.

#### 305. Structural Analysis I

Fall, Winter. 3(3-0) MMM 211.

Stability and determinacy; linearity. Plane trusses; shear and bending in beams and frames. Virtual work calculation of forces and displacements in statically-determinate plane structures.

#### 306. Structural Analysis II

Winter, Spring. 3(3-0) C E 305.

Beam-columns. Flexibility method for plane structures. Kinematic indeterminacy. Stiffness method for plane structures. Moment distribution method. Design loading of structures.

#### 308. Engineering Materials I

Fall, Winter, Spring. 4(3-3) MMM 211 or concurrently.

Structure; composition; physical, mechanical and rheological properties of non-metallic construction materials. Emphasis on aggregates, asphalt, inorganic cements, concrete, and wood.